

## Module 7: Solved Problems

1. A thin-walled concentric tube heat exchanger of 0.19-m length is to be used to heat deionized water from 40 to 60°C at a flow rate of 5 kg/s. the deionized water flows through the inner tube of 30-mm diameter while hot process water at 95°C flows in the annulus formed with the outer tube of 60-mm diameter. The thermo physical properties of the fluids are:

	DEIONIZED WATER	PROCESS WATER
$\rho(\text{kg/m}^3)$	982.3	967.1
$c_p(\text{J/kg.K})$	4181	4197
$k(\text{W/m.K})$	0.643	0.673
$\mu(\text{N.s/m}^2)*10^6$	548	324
$Pr$	3.56	2.02

Considering a parallel-flow configuration of the exchanger, determine the minimum flow rate required for the hot process water.

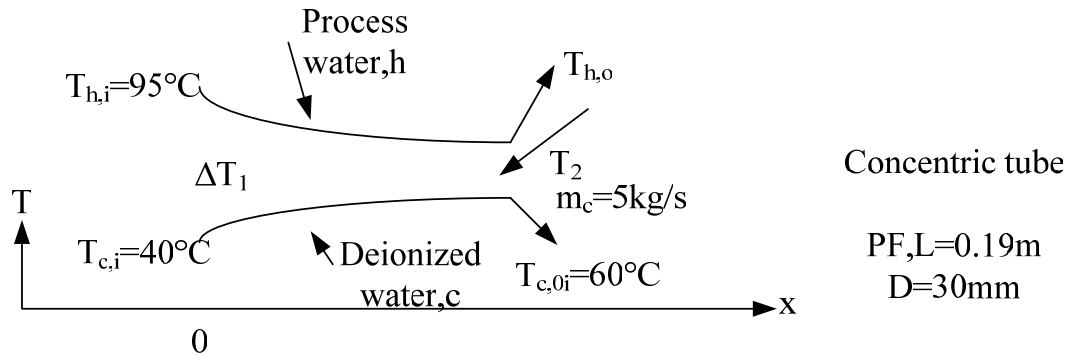
Determine the overall heat transfer coefficient required for the conditions of part a.

Considering a counter flow configuration, determine the minimum flow rate required for the hot process water. What is the effectiveness of the exchanger for this situation?

Known: Thin-walled concentric tube, Parallel flow heat exchanger of prescribed diameter and length with process and deionized water. Inlet and outlet temperatures and flow rate of desired water. Inlet temperature and outlet temperature and flow rate of deionized water. Inlet temperature of process water.

Find: (1) minimum flow rate required for the hot process water, (b) required overall heat transfer coefficient and whether it is possible to accomplish this heating, and (c) for CF arrangements minimum process water flow required and the effectiveness?

Schematic:



Assumptions: (1) Negligible heat loss to surroundings, (2) Negligible kinetic and potential energy changes.

Analysis: (a) from overall energy balances,

$$q = (\dot{m}c)_h (T_{h,i} - T_{h,o}) = (\dot{m}c)_c (T_{c,o} - T_{c,i})$$

For a fixed term  $T_{h,i}$ ,  $(\dot{m})_h$  will be a minimum when  $T_{h,o}$  is a minimum. With the parallel flow configuration, this requires that  $T_{h,o} = T_{c,o} = 60^\circ\text{C}$ . Hence,

$$\dot{m}h, \min = \frac{(\dot{m}c)_c (T_{c,o} - T_{c,i})}{c_h (T_{h,i} - T_{h,o})} = \frac{5 \text{ kg/s} \times 4181 \text{ J/kg}\cdot\text{K} (60 - 40)^\circ\text{C}}{4197 \text{ J/kg}\cdot\text{K} (95 - 60)^\circ\text{C}} = 2.85 \text{ kg/s}$$

(b) From the rate equation and the log mean temperature relation,

$$q = UA\Delta T_{lm,PF}$$

$$\Delta T_{lm,PF} = \frac{\Delta T_1 - \Delta T_2}{\ln\left(\frac{\Delta T_1}{\Delta T_2}\right)}$$

And since  $\Delta T_2=0$ ,  $\Delta T_{lm}=0$  so that  $UA=\infty$ . Since  $A=\pi DL$  is finite,  $U$  must be extremely large. Hence, the heating cannot be accomplished with this arrangement.

(c) With the CF arrangements  $\dot{m}_h$  will be a minimum when  $T_{ho}$  is a minimum. This requires that  $T_{h,o}$  is a minimum. This requires that  $T_{h,o}$  is a minimum. This requires that  $T_{h,o}=T_{c,i}=40^\circ\text{C}$ . Hence, from the overall energy balance,

$$\dot{m} = \frac{5\text{kg/s} \times 4181\text{J/kg}\cdot\text{K}(60 - 40)\text{K}}{4197\text{J/kg}\cdot\text{K}(95 - 40)\text{K}} = 1.81\text{kg/s}$$

For this condition,  $C_{\min}=C_h$  which is cooled from  $T_{h,i}$  to  $T_{c,i}$ , hence  $\varepsilon=1$

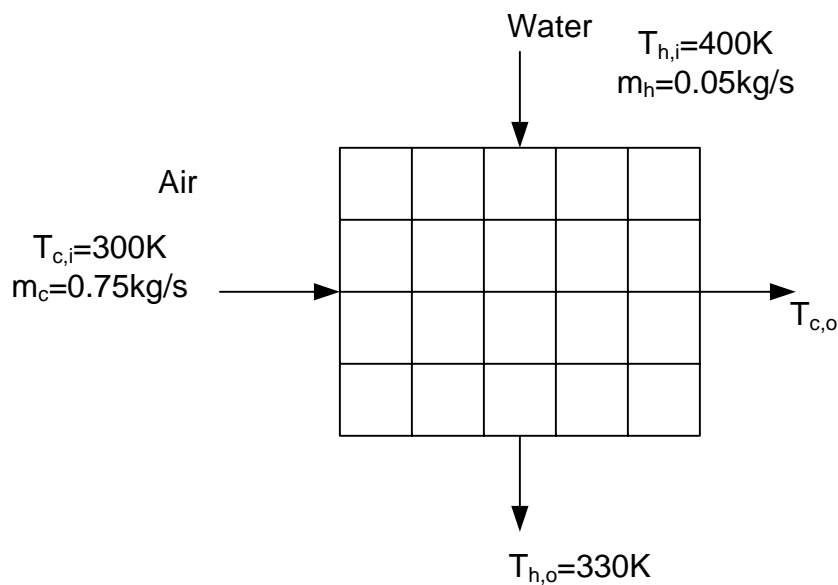
Comments: For the counter flow arrangement, the heat exchanger must be infinitely long.

2. An automobile radiator may be viewed as a cross-flow heat exchanger with both fluids unmixed. Water, which has flow rate of  $0.05\text{kg/s}$ , enters the radiator at  $400\text{K}$  and is to leave at  $330\text{K}$ . The water is cooled by air which enters at  $0.75\text{kg/s}$  and  $300\text{K}$ . If the overall heat transfer coefficient is  $200\text{W/m}^2\cdot\text{K}$ , what is the required heat transfer surface area?

Known: flow rate and inlet temperature for automobile radiator. Overall heat transfer coefficient.

Find: Area required to achieve a prescribed outlet temperature.

Schematic:



Assumptions: (1) Negligible heat loss to surroundings and kinetic and potential energy changes, (2) Constant properties.

Analysis: The required heat transfer rate is

$$q = (\dot{m}c)_h (T_{h,i} - T_{h,o}) = 0.05\text{kg/s}(4209\text{J/kg}\cdot\text{K})70\text{K} = 14,732\text{W}$$

Using the  $\varepsilon$ -NTU method,

$$C_{\min} = C_h = 210.45W / K$$

$$C_{\max} = C_c = 755.25W / K,$$

$$\text{hence, } C_{\min} / C_{\max} (T_{h,i} - T_{c,i}) = 210.45W / K (100K) = 21,045W$$

*and*

$$\varepsilon = q / q_{\max} = 14,732W / 21,045W = 0.700$$

From figure,  $NTU \approx 1.5$ , hence

$$A = NTU(C_{\min} / U) = 1.5 \times 210.45W / K (200W / m^2 \cdot K) = 1.58m^2$$

Comments: (1) the air outlet temperature is

$$T_{c,o} = T_{c,i} + q / C_c = 300K + (14,732W / 755.25W / K) = 319.5K$$

(2) Using the LMTD approach,  $\Delta T_{lm} = 51.2$  K,  $R = 0.279$  and  $P = 0.7$ .  
Hence from fig  $F \approx 0.95$  and

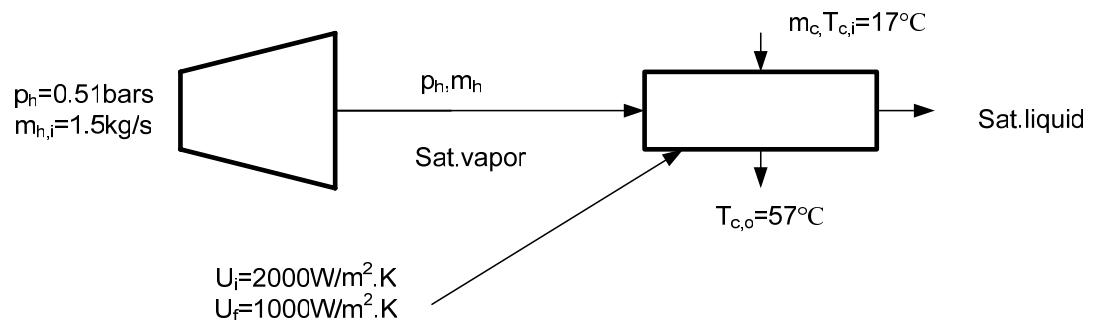
$$A = q / F U \Delta T_{lm} = (14,732W) / [0.95(200W / m^2 \cdot K) 51.2K] = 1.51m^2.$$

3. Saturated water vapor leaves a steam turbine at a flow rate of  $1.5\text{kg/s}$  and a pressure of  $0.51\text{ bars}$ . The vapor is to be completely condensed to saturated liquid in a shell-and-tube heat exchanger which uses city water as the cold fluid. The water enters the thin-walled tubes at  $17^\circ\text{C}$  and is to leave at  $57^\circ\text{C}$ . assuming an overall heat transfer coefficient of  $200\text{W/m}^2\cdot\text{K}$ , determine the required heat exchanger surface area and the water flow rate. After extended operation, fouling causes the overall heat transfer coefficient to decrease to  $100\text{W/m}^2\cdot\text{K}$ , and to completely condense the vapor, there must be an attendant reduction in the vapor flow rate. For the same water inlet temperature and flow rate, what is the new vapor flow rate required for complete condensation?

Known: Pressure and initial flow rate of water vapor. Water inlet and outlet temperatures. Initial and final overall heat transfer coefficients.

Find: (a) Surface area for initial  $U$  and water flow rate, (b) Vapour flow rate for final  $U$ .

Schematic:



Assumptions: (1) Negligible heat loss to surroundings, (2) Negligible wall conduction resistance.

Properties: Table for  
sat. Water:

$$(\bar{T}_c = 310\text{K}) : c_{p,c} = 4178\text{J/kg.K}; (p = 0.51\text{ bars}) : T_{\text{sat}} = 355\text{K}, h_{fg} = 2304\text{kJ/kg}.$$

Analysis: (a) The required heat transfer rate is

$$q = \dot{m}_h h_{fg} = 1.5\text{kg/s}(2.304 \times 10^6\text{ J/kg}) = 3.46 \times 10^6\text{ W}$$

And the corresponding heat capacity rate for the water is

$$C_c = C_{\min} = q / (T_{c,o} - T_{c,i}) = 3.48 \times 10^6\text{ W} / 40\text{K} = 86,400\text{ W} / \text{K}$$

$$\text{hence, } \varepsilon = q / (C_{\min} [T_{h,i} - T_{c,i}]) = 3.46 \times 10^6\text{ W} / 86,400\text{ W} / \text{K}(65\text{K}) = 0.62$$

$$\text{since } C_{\min} / C_{\max} = 0,$$

$$\text{NTU} = -\ln(1 - \varepsilon) = -\ln(1 - 0.62) = 0.97$$

And

$$A = \text{NTU}(C_{\min} / U) = 0.97(86,400\text{ W} / \text{K} / 2000\text{ W} / \text{m}^2\text{.K}) = 41.9\text{m}^2$$

$$\dot{m}_c = C_c / c_{p,c} = 86,400\text{ W} / \text{K} / 4178\text{J/kg.K} = 20.7\text{kg/s}$$

(b) using the final overall heat transfer coefficient, find

$$\text{Since } C_{\min} / C_{\max} = 0,$$

$$\varepsilon = 1 - \exp(-NTU) = 1 - \exp(-0.485) = 0.384$$

$$\text{hence, } q = \varepsilon C_{\min} (T_{h,i} - T_{c,i}) = 0.384(886,400W / K)65K = 2.16106W$$

$$\dot{m}_h = q / h_{fg} = 2.16 \times 10^6 W / 2.304 \times 10^6 J / kg = 0.936kg / s$$

Comments: The significant reduction (38%) in  $\dot{m}_h$  represents a significant loss in turbine power. Periodic cleaning of condenser surfaces should be employed to minimize the adverse effects of fouling.

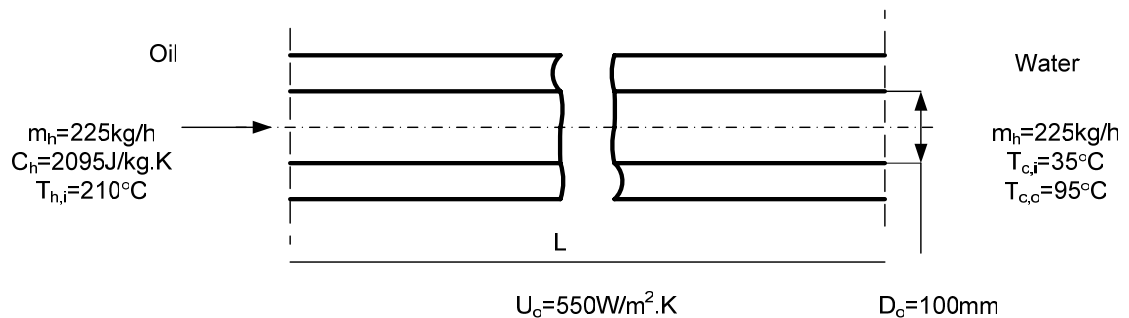


4. Water at 225 kg/h is to be heated from 35 to 95°C by means of a concentric tube heat exchanger. Oil at 225kg/h and 210°C, with a specific heat of 2095 J/kg.K, is to be used as the hot fluid, If the overall heat transfer coefficient based on the outer diameter of the inner tube is 550W/m<sup>2</sup>.K, determine the length of the exchanger if the outer diameters is 100mm.

Known: Concentric tube heat exchanger.

Find: Length of the exchanger

Schematic:



Assumptions: (1) Negligible heat loss to surroundings, (2) Negligible kinetic and potential energy changes, (3) Constant properties.

Properties: Table for Water:

$$(\bar{T}_c = (35 + 95)^\circ\text{C} / 2 = 338\text{K}) : c_{p,c} = 4188 \text{ J / kg.K}$$

Analysis: From rate equation with  $A_o = \pi D_o L$ ,  $L = q / U_o D_o \Delta T \ell_m$

The heat rate,  $q$ , can be evaluated from an energy balance on the cold fluid,

$$q = \dot{m}_c c_c (T_{c,o} - T_{c,i}) = \frac{225 \text{ kg / h}}{3600 \text{ s / h}} \times 4188 \text{ J / kg.K} (95 - 35) \text{ K} = 15,705 \text{ W}$$

In order to evaluate  $\Delta T_{lm}$ , we need to know whether the exchanger is operating in CF or PF. From an energy balance on the hot fluid, find

$$T_{h,o} = T_{h,i} - q / \dot{m}_h c_h = 210^\circ C - 15,705W / \frac{225kg/h}{3600s/h} \times 2095 \frac{J}{kg.K} = 90.1^\circ C$$

Since  $T_{h,o} < T_{c,o}$  it follows that HXer operation must be CF. From eq. for log mean temperature difference,

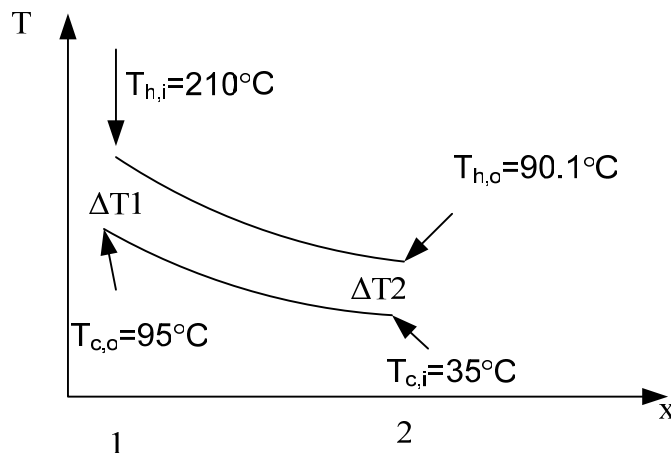
$$\Delta T_{lm,CF} = \frac{\Delta T_1 - \Delta T_2}{\ln(\Delta T_1 / \Delta T_2)} = \frac{(210 - 95) - (90.1 - 35)}{\ln(115 / 55.1)}^\circ C = 81.5^\circ C$$

Substituting numerical values, the HXer length is

$$L = 15,705W / 550W / m^2 .K \pi(0.10m) \times 81.4K = 1.12m$$

Comments: The  $\epsilon$ -NTU method could also be used. It would be necessary to perform the hot fluid energy balance to determine if CF operation existed. The capacity rate is  $C_{min}/C_{max}=0.50$ . From eq. for effectiveness, and from with  $q$  evaluated from an energy balance on the hot fluid,

$$\epsilon = \frac{T_{h,i} - T_{h,o}}{T_{h,i} - T_{c,i}} = \frac{210 - 90.1}{210 - 35} = 0.69$$



From fig, find  $NTU \approx 1.5$  giving

$$L = NTU \cdot C_{\min} / U_o \pi D_o \approx 1.5 \times 130.94 \frac{W}{K} 550 \frac{W}{m^2 \cdot K} \cdot \pi(0.10m) \approx 1.14m$$

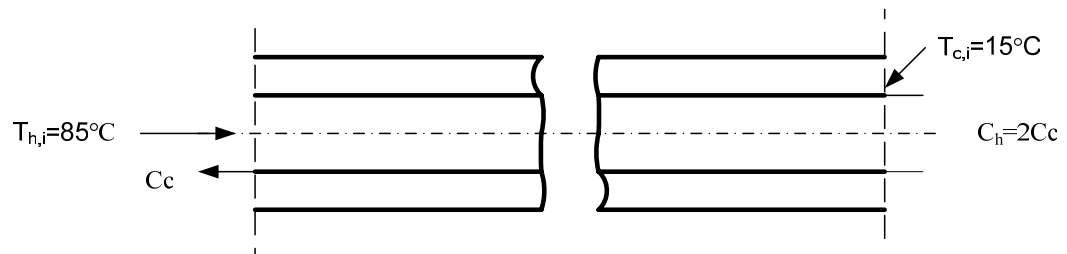
Note the good agreement by both methods.

5. Consider a very long, concentric tube heat exchanger having hot and cold water inlet temperatures of 85 and 15°C. The flow rate of the hot water is twice that of the cold water. Assuming equivalent hot and cold water specific heats; determine the hot water outlet temperature for the following modes of operation (a) Counter flow, (b) Parallel flow.

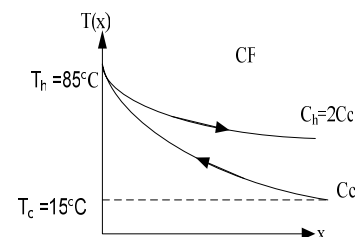
Known: A very long, concentric tube heat exchanger having hot and cold water inlet temperatures of 85 and 15°C, respectively: flow rate of the hot water is twice that of the cold water.

Find: outlet temperatures for counter flow and parallel flow operations.

Schematic:



Assumptions: (1) equivalent hot and cold water specific heats, (2) Negligible Kinetic and potential energy changes, (3) No heat loss to surroundings.



Analysis: the heat rate for a concentric tube Heat exchanger with very large surface area Operating in the counter flow mode is

$$q = q_{\max} = C_{\min} (T_{h,i} - T_{c,i})$$

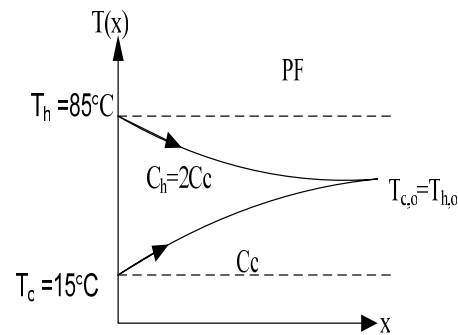
Combining the above relation and rearranging, find

$$T_{h,o} = -\frac{C_{\min}}{C_h} (T_{h,i} - T_{c,i}) + T_{h,i} = -\frac{C_c}{C_h} (T_{h,i} - T_{c,i}) + T_{h,i}$$

Substituting numerical values

$$T_{h,o} = -\frac{1}{2} (85 - 15)^\circ C + 85^\circ C = 50^\circ C$$

For parallel flow operation, the hot and cold outlet temperatures will be equal; that is  $T_{c,o} = T_{h,o}$ . Hence



$$C_c (T_{c,o} - T_{c,i}) = C_h (T_{h,i} - T_{h,o})$$

Setting  $T_{c,o} = T_{h,o}$  and rearranging

$$T_{h,o} = \left[ T_{h,i} + \frac{C_c}{C_h} T_{c,i} \right] / \left[ 1 + \frac{C_c}{C_h} \right]$$

$$T_{h,o} = \left[ 85 + \frac{1}{2} \times 15 \right]^\circ C / \left[ 1 + \frac{1}{2} \right] = 61.7^\circ C$$

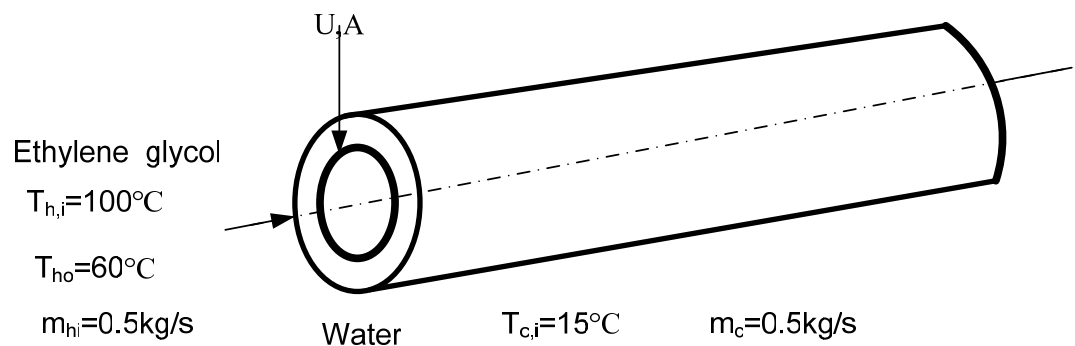
Comments: Note that while  $\varepsilon = 1$  for CF operation, for PF operation find  $\varepsilon = q/q_{\max} = 0.67$ .

6. A concentric tube heat exchanger uses water, which is available at 15C, to cool ethylene glycol from 100 to 60C. The water and glycol flow rates are each 0.5 kg/s. What are the maximum possible heat transfer rate and effectiveness of the exchanger? Which is preferred, a parallel –flow or counter flow mode of operation?

Known: Inlet temperatures and flow rate for a concentric tube heat exchanger.

Find: (a) Maximum possible heat transfer rate and effectiveness, (b) Proffered mode of operation.

Schematic:



Assumptions: (1) Steady-state operation, (2) Negligible KE and PE changes, (3) Negligible heat loss to surroundings, (4) Fixed overall heat transfer and coefficient.

Properties: Table: Ethylene glycol ( $\bar{T}_{in} = 80^\circ\text{C}$ );  $c_p = 2650 \text{ J/kg}\cdot\text{K}$ ;

**Water ( $\bar{T}_m \approx 30^\circ\text{C}$ ):  $c_p = 4178 \text{ J/kg}\cdot\text{K}$**

Analysis: (a) Using the  $\epsilon$ -NTU method, find

$$C_{\min} = C_h = \dot{m}_h c_{p,h} = (0.5 \text{ kg/s})(2650 \text{ J/kg.K}) = 1325 \text{ W/K}$$

$$q_{\max} = C_{\min} (T_{h,i} - T_{c,i}) = (1325 \text{ W/K})(100 - 15)^\circ\text{C} = 1.13 \times 10^5 \text{ W}$$

$$q = \dot{m}_h c_{p,h} (T_{h,i} - T_{c,i}) = 0.5 \text{ kg/s}(2650 \text{ J/kg.K})(100 - 60)^\circ\text{C} = 0.53 \times 10^5 \text{ W}$$

$$\varepsilon = q / q_{\max} = 0.53 \times 10^5 / 1.13 \times 10^5 = 0.47$$

(b)

$$T_{c,o} = T_{c,i} + \frac{q}{\dot{m}_c c_{p,c}} = 15^\circ\text{C} + \frac{0.53 \times 10^5}{0.5 \text{ kg/s} \times 4178 \text{ J/kg.K}} = 40.4^\circ\text{C}$$

Since  $T_{c,o} < T_{h,o}$ , a parallel flow mode of operation is possible.

However, with  $(C_{\min}/C_{\max}) = (\dot{m}_h c_{p,h} / \dot{m}_c c_{p,c}) = 0.63$ ,

From fig  $(\text{NTU})_{\text{PF}} \approx 0.95$ ,  $(\text{NTU})_{\text{CF}} \approx 0.75$

Hence

$$(A_{\text{CF}}/A_{\text{PF}}) = (\text{NTU})_{\text{CF}} / (\text{NTU})_{\text{PF}} \approx (0.75/0.95) = 0.79$$

Because of the reduced size requirement, hence capital investment, the counter flow mode of operation is proffered.